



UNIVERSITÀ  
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Gestionale, Informatica, Meccanica (DICGIM)



**REAPower**

[www.reapower.eu](http://www.reapower.eu)

# REAPower: Use of Desalination Brine for Power Production through Reverse Electrodialysis

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International Conference *WIN4Life*  
Tinos Island (Greece) – 19<sup>th</sup>-21<sup>st</sup> September 2013

# The REAPower Project

## Main facts:



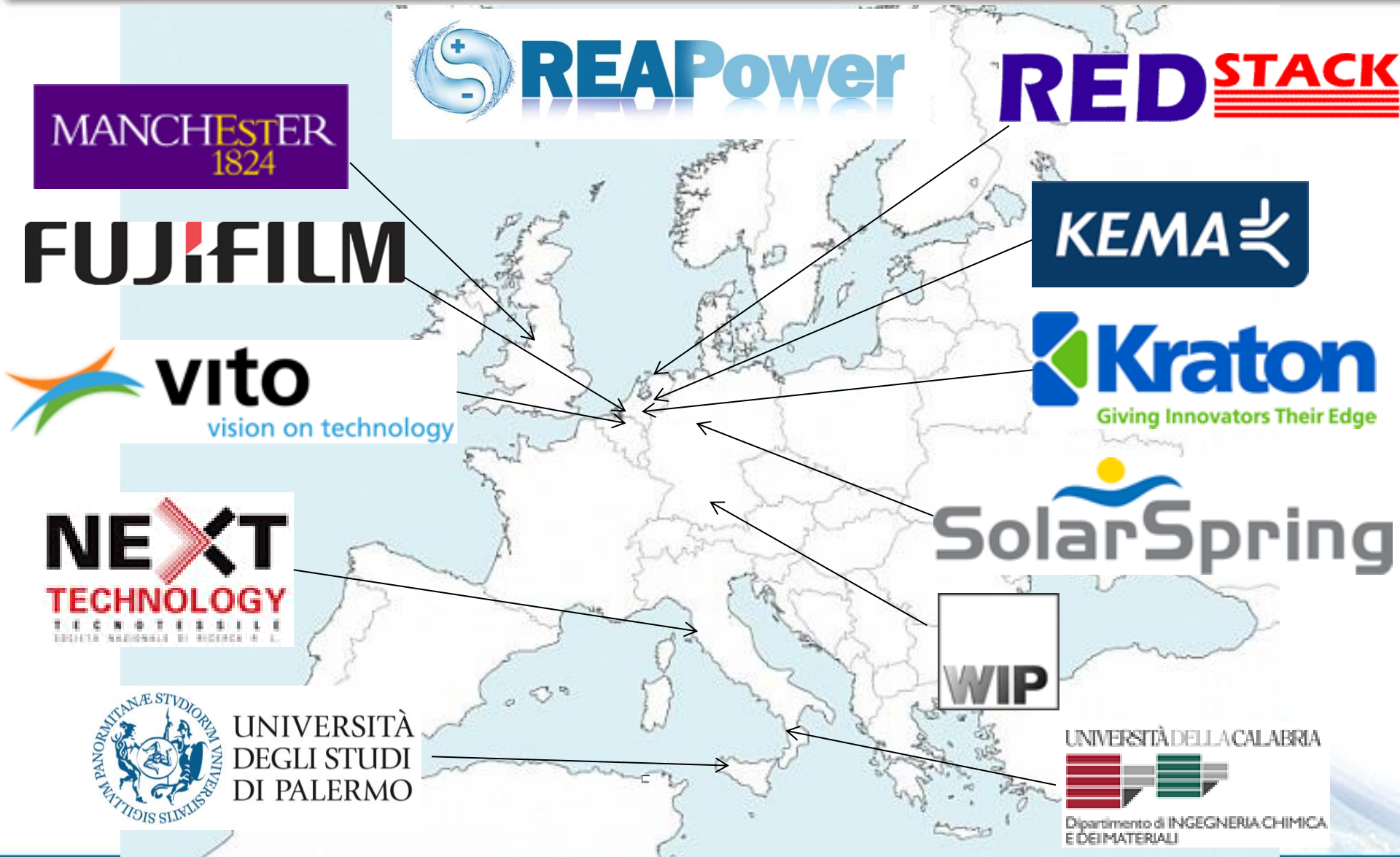
- Project title: *Reverse Electrodialysis Alternative Power Production*
- Call identifier: FP7-ENERGY-2010-FET (Future Emerging Technologies for Energy Applications)
- Starting date: 1 October 2010
- Closing date: 30 September 2014

# The Future

of sustainable energy production

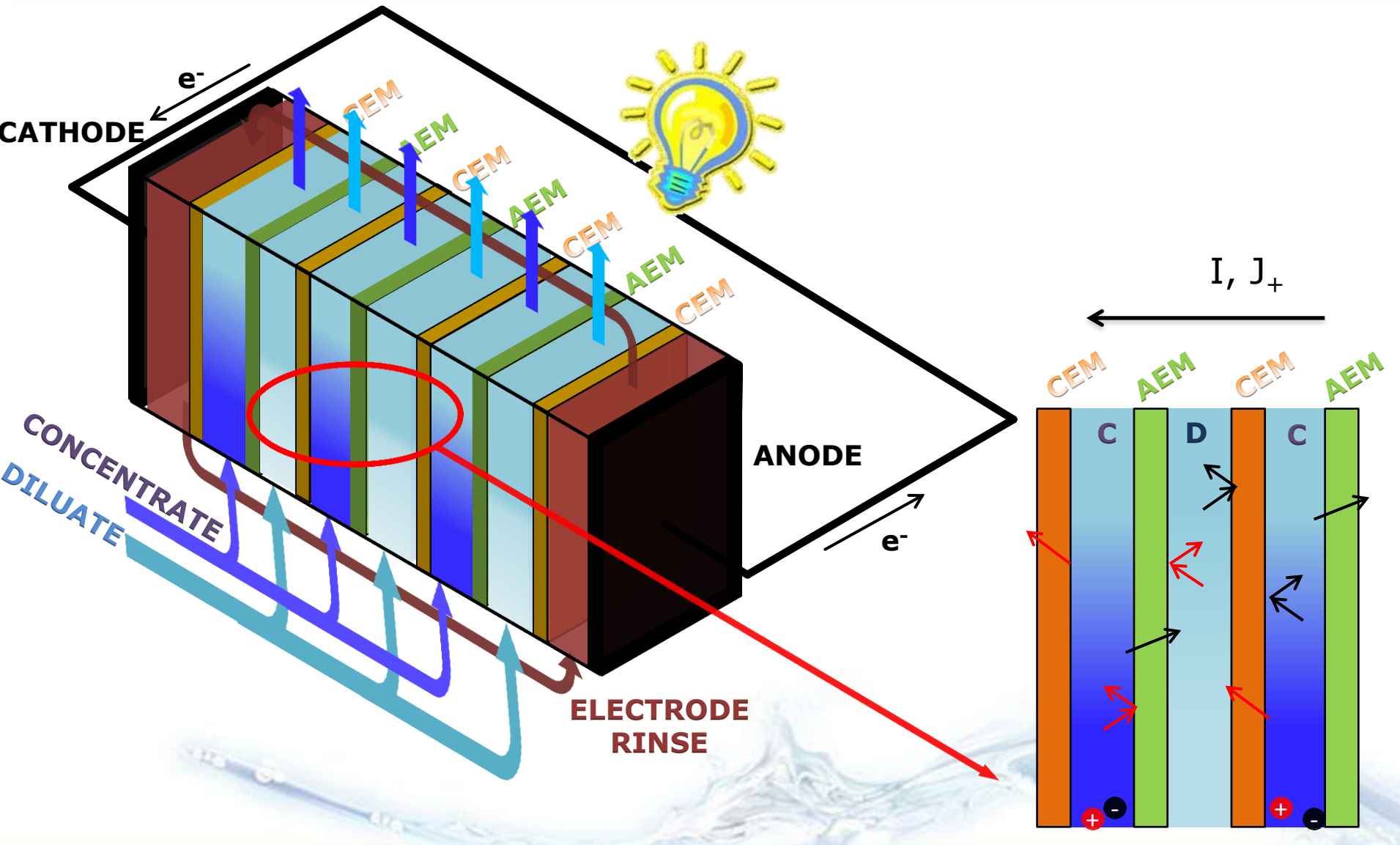


# The REAPower Project Consortium





# The Reverse Electrodialysis technology





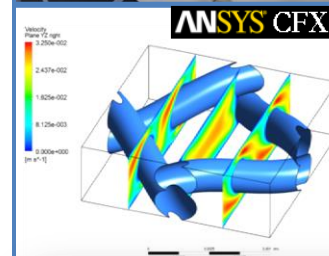
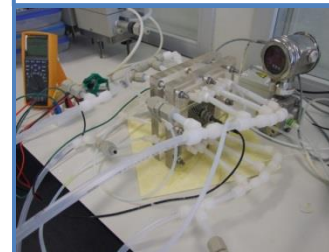
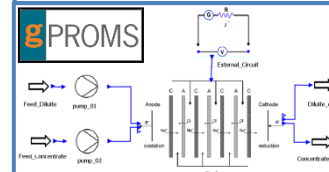
# The REAPower Project

## The idea

to produce **energy from salinity gradients** generated by **sea/brackish water** and **ultra-concentrated brines**

## R&D strategy

- ✓ Development of new **Ion Exchange Membranes** for highly concentrated solutions
- ✓ Selection of best conditions for **redox couple/stack design**
- ✓ Wide **experimental investigation** on lab-scale stack
- ✓ Development/validation of a **predictive modelling tool**
- ✓ **Economic analysis** and process sustainability on large scale





# Improvements in membranes development

Increased **permselectivity**



✓ Permselectivity has achieved values of **65% for AEM** and **up to 90% for CEM** when **in contact with** almost saturated **brine**

Reduced **membrane resistance**



✓ Membrane specific resistance reduced to **1.5-2.5  $\Omega \cdot \text{cm}^2$**  aiming at a 3-5 folds reduction in the next months

# Electrochemical aspects and stack design

Redox couples selection



Investigated redox couples under different conditions:

- ✓  $\text{FeCl}_3/\text{FeCl}_2$
- ✓  $\text{K}_3\text{Fe}(\text{CN})_6/\text{K}_4\text{Fe}(\text{CN})_6$
- ✓ Fe(III)-EDTA/Fe(II)-EDTA

New stack design



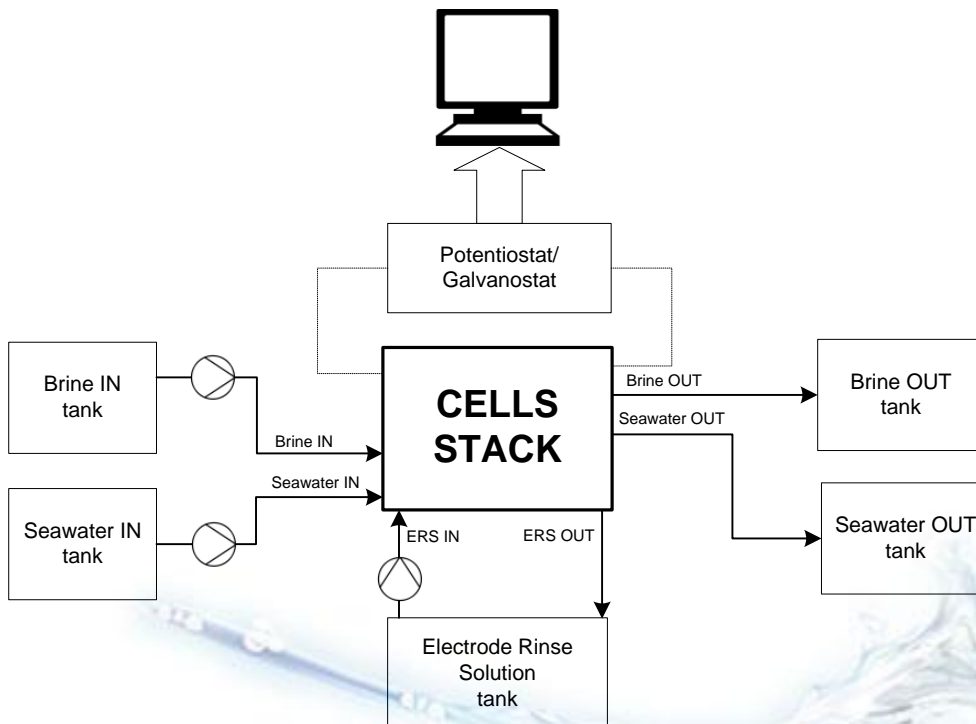
- ✓ Two different **stack geometries** already designed, constructed and tested
- ✓ Currently available for the consortium



# Experimental investigation on a lab-scale unit

## Experimental conditions already investigated:

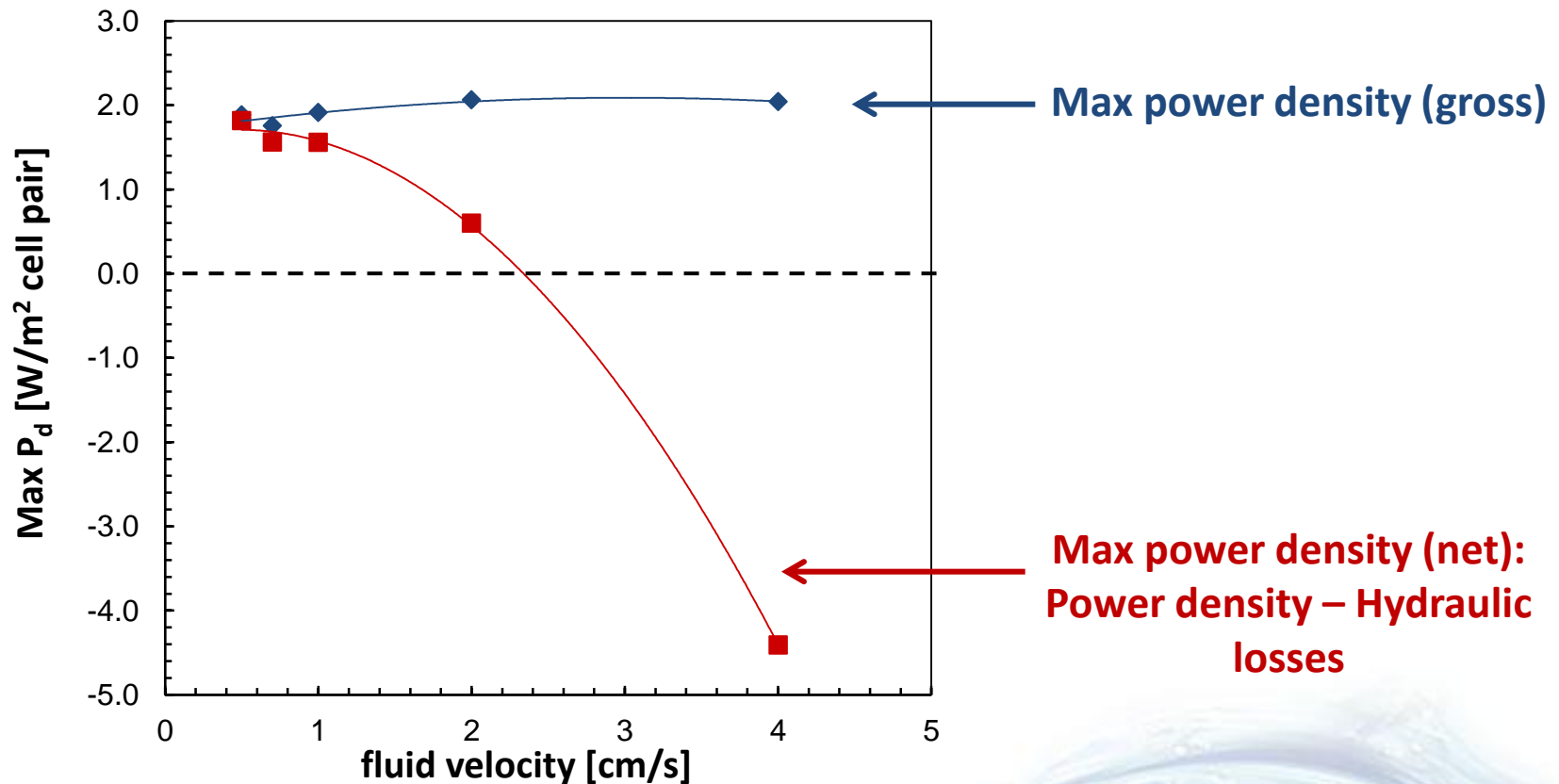
- ✓ fluid velocity (0.1 – 4 cm/s)
- ✓ feed temperature (20 – 40 °C)
- ✓ number of cell pairs (5 – 50)
- ✓ concentration of redox couple (0.1 – 0.3 M of  $K_3Fe(CN)_6/K_4Fe(CN)_6$ )
- ✓ salt concentration of both solutions.





# Experimental investigation on a lab-scale unit

## Effect of fluid velocity on power output

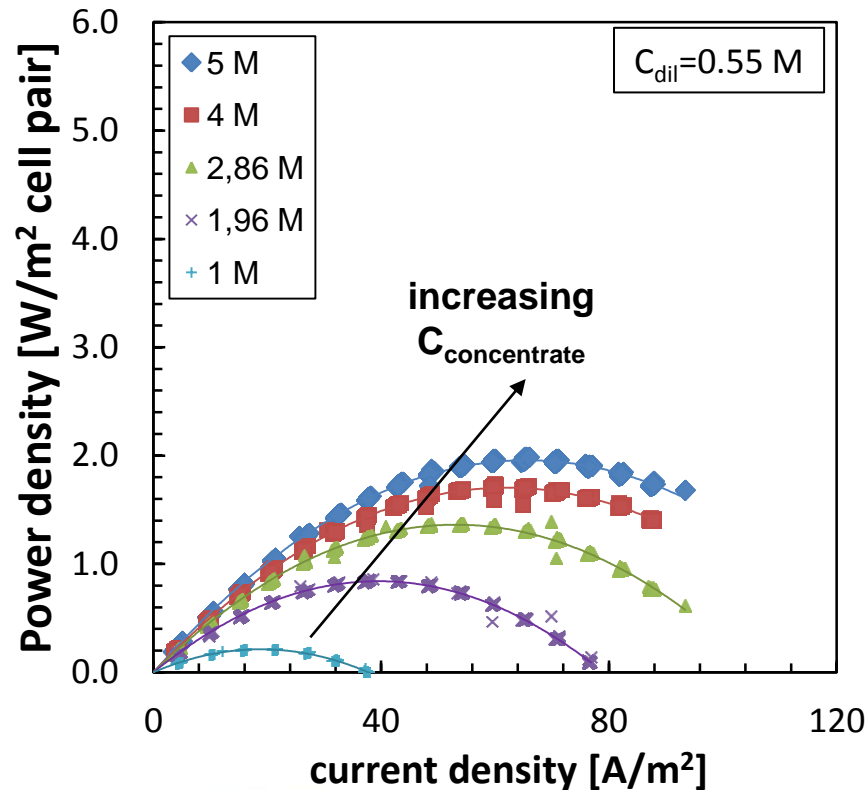


Stack equipped with 50 cell pairs, Fujifilm membranes, Deukum 270  $\mu\text{m}$  spacers. Brine solution: 5 M NaCl, seawater: 0.5 M NaCl.  $T=20^\circ\text{C}$ . Electrode rinse solution: 0.1 M  $\text{K}_3\text{Fe}(\text{CN})_6 / \text{K}_4\text{Fe}(\text{CN})_6 \cdot 3\text{H}_2\text{O} + 2.5\text{M NaCl}$ .

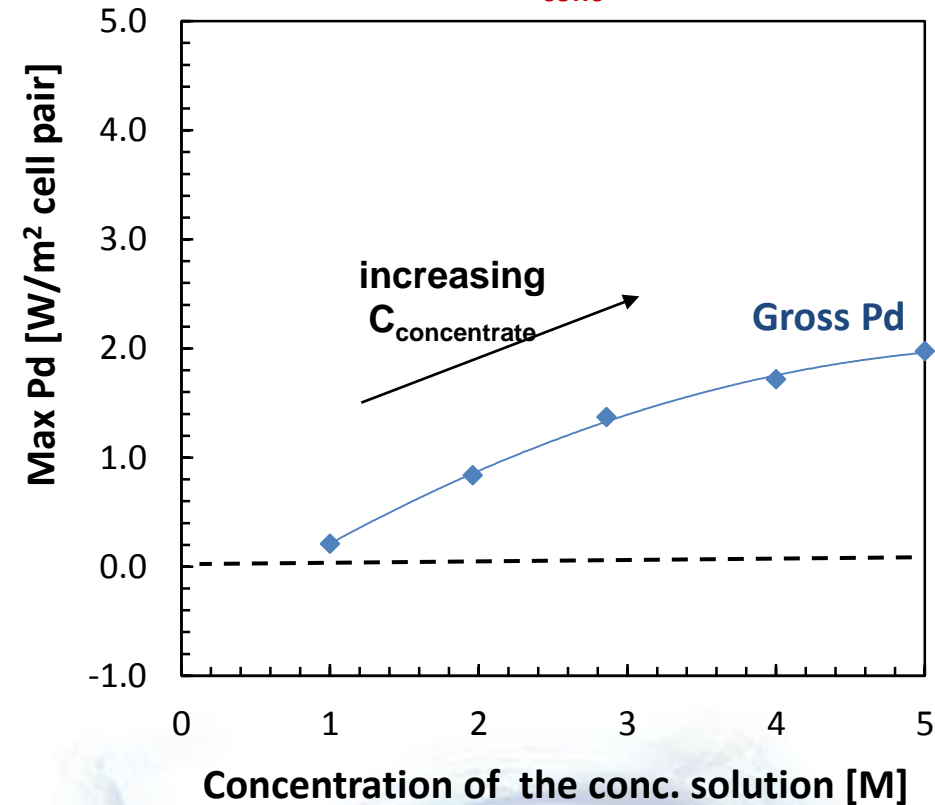
# Experimental investigation on a lab-scale unit

## Effect of the concentration of the concentrated solution (1 ÷ 5 M)

Power density vs.  
current density



Maximum power density  
vs.  $C_{\text{conc}}$

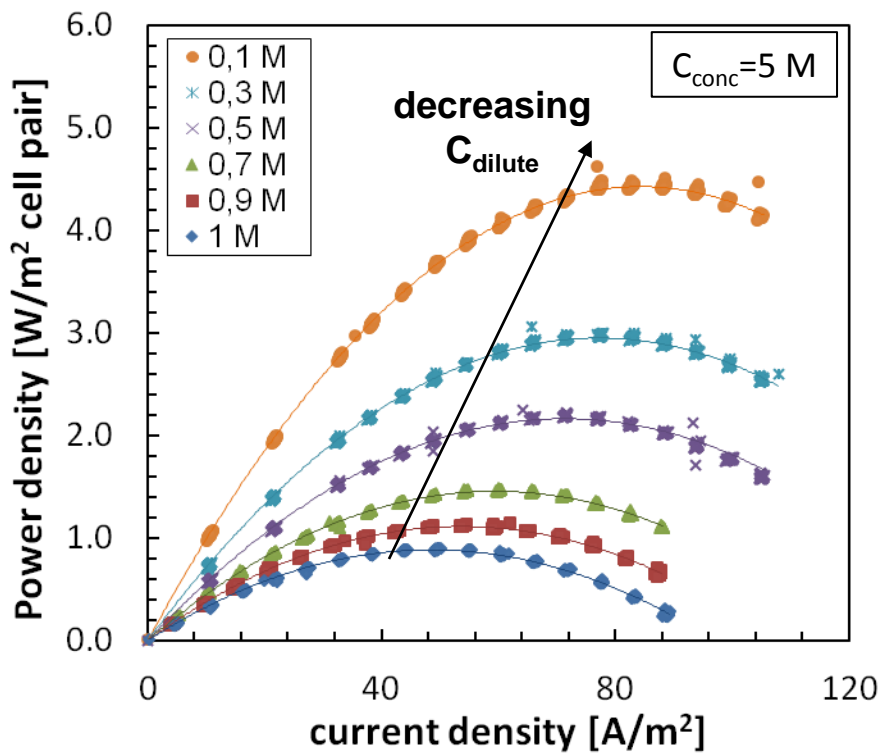


Stack equipped with 50 cell pairs, Fujifilm membranes, Deukum 270  $\mu\text{m}$  spacers . Seawater: 0.55 M NaCl.  $T=20^\circ \text{C}$ . Fluid velocity: 1 cm/s.  
Electrode rinse solution: 0.1 M  $\text{K}_3\text{Fe}(\text{CN})_6 / \text{K}_4\text{Fe}(\text{CN})_6 \cdot 3\text{H}_2\text{O} + 2.5 \text{ M NaCl}$ .

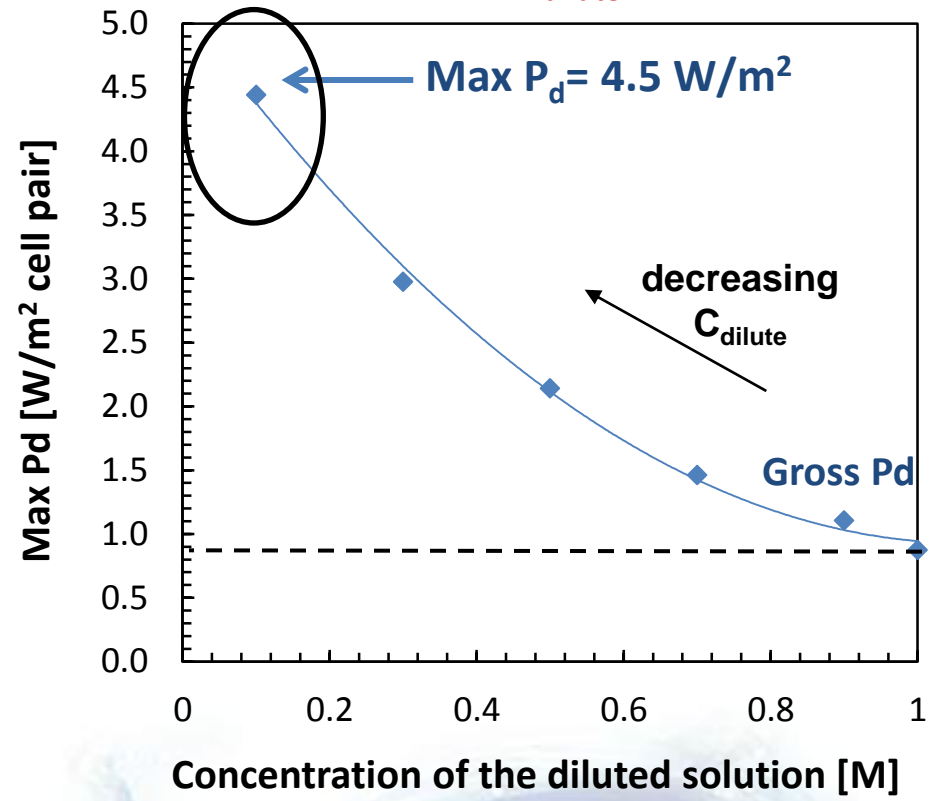
# Experimental investigation on a lab-scale unit

## Effect of the concentration of the diluted solution (0.1 ÷ 1 M)

### Power density vs. current density



### Maximum power density vs. $C_{dilute}$



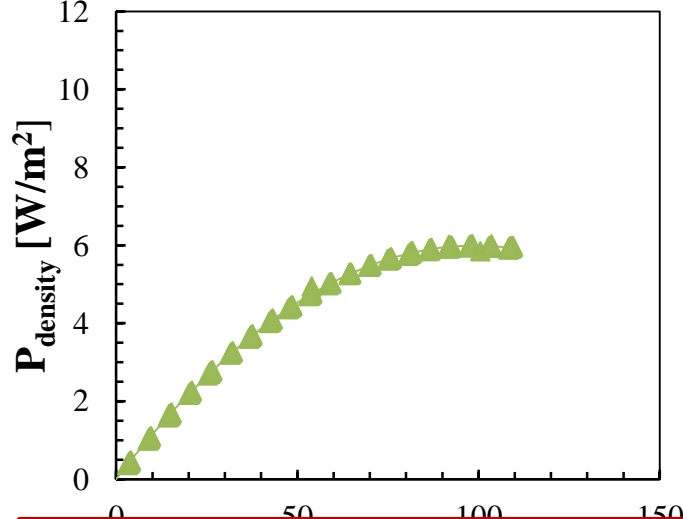
Stack equipped with 50 cell pairs, Fujifilm membranes, Deukum 270  $\mu\text{m}$  spacers . Brine: 5 M NaCl.  $T=20^\circ \text{ C}$ . Fluid velocity: 1 cm/s. Electrode rinse solution: 0.1 M  $\text{K}_3\text{Fe}(\text{CN})_6 / \text{K}_4\text{Fe}(\text{CN})_6 \cdot 3\text{H}_2\text{O} + 2.5 \text{ M NaCl}$ .

# Experimental investigation on a lab-scale unit

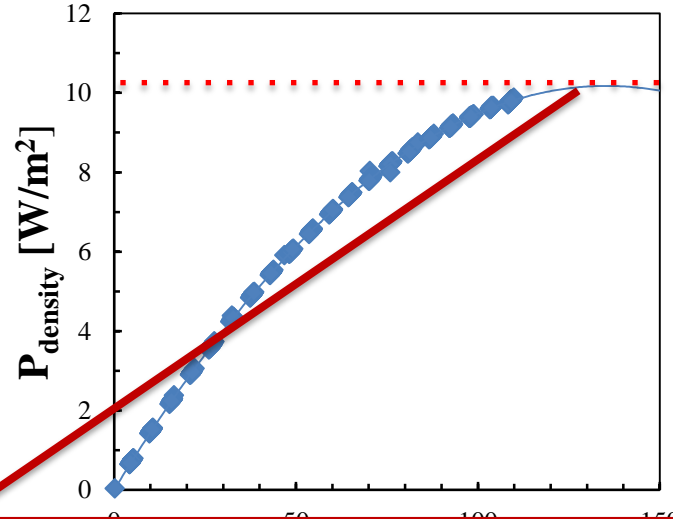
## MAX power output conditions:

4cm/s, T = 40°C & brackish water diluate (0.1M)

Thick membranes (120 μm)  
(provided by Fujifilm)



Thin membranes (20-30mm)  
(provided by Fumatech)

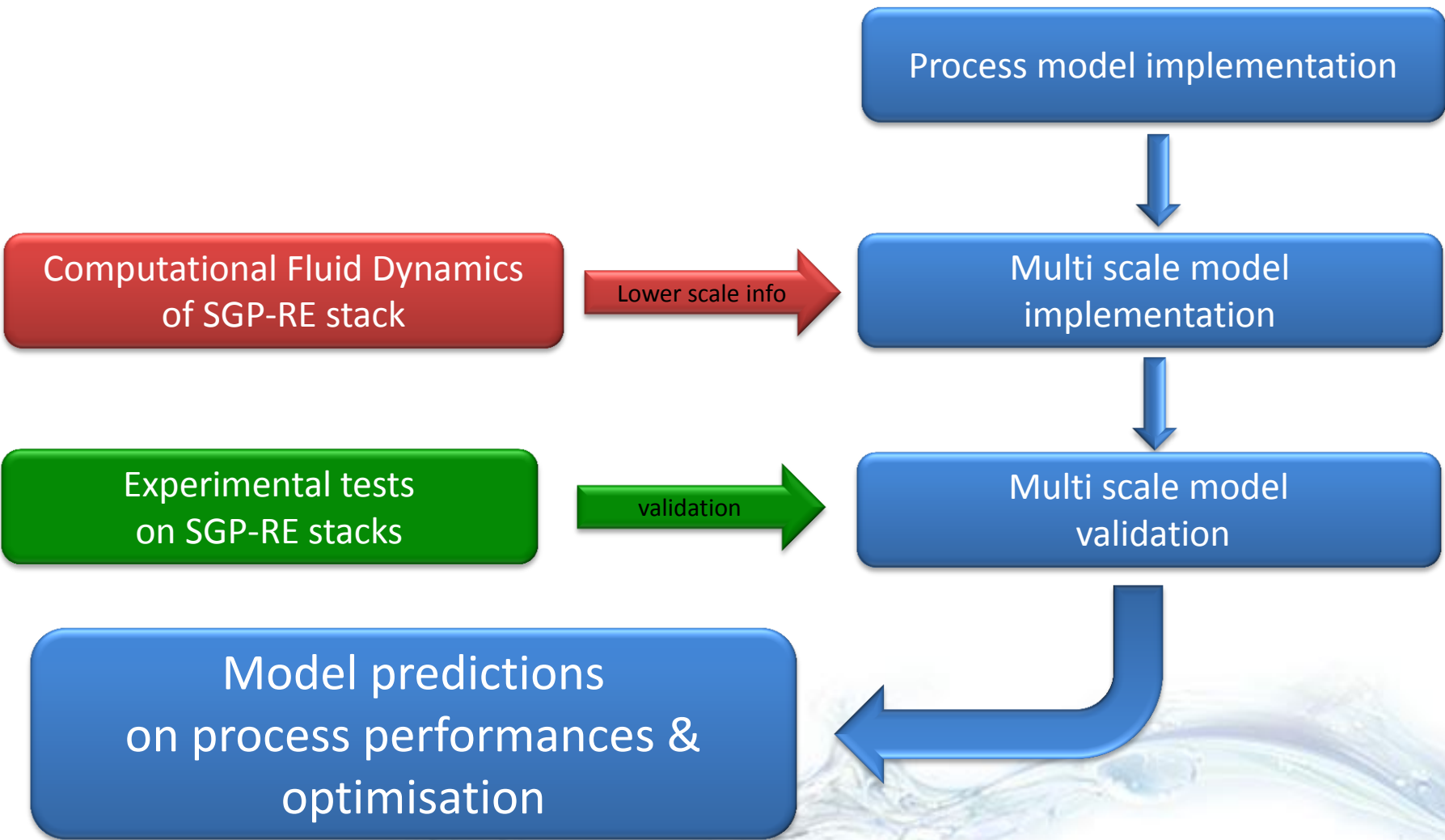


Power density between **15 and 20 W/m<sup>2</sup>** can be expected with larger number of cell pairs, i.e. reducing the effect of **blank resistance**

50 ce

# CFD modelling and process simulation

## Multi-Scale Modelling approach:

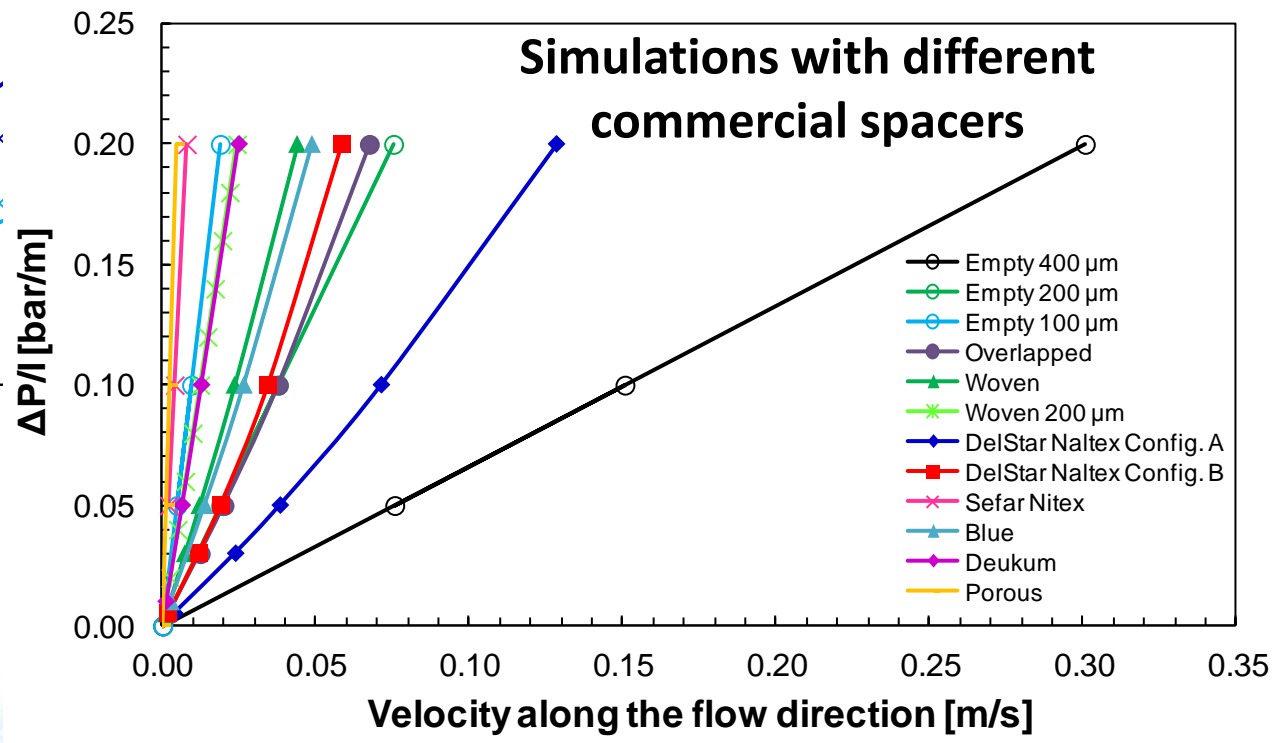
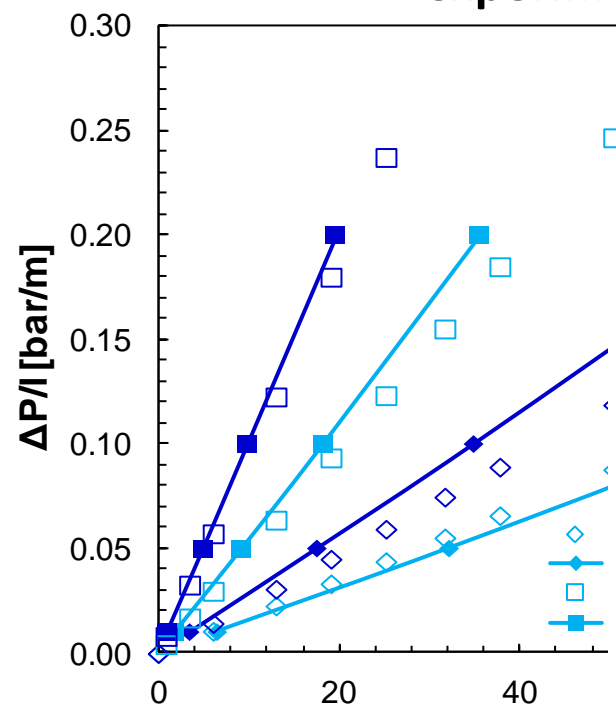






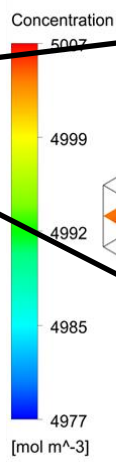
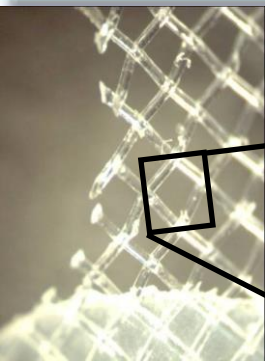
# CFD Modelling: prediction of pressure drops

## Model validation with experimental results

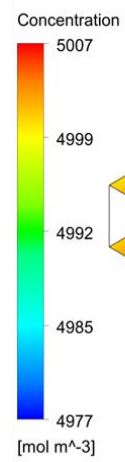
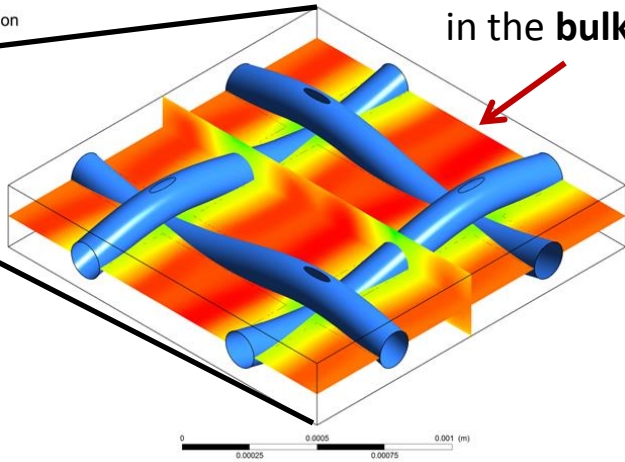




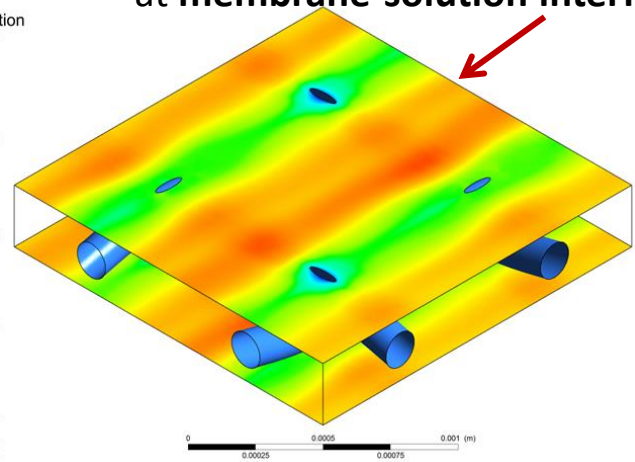
# CFD Modelling: prediction of polarisation phenomena



salt concentration map in the **bulk**

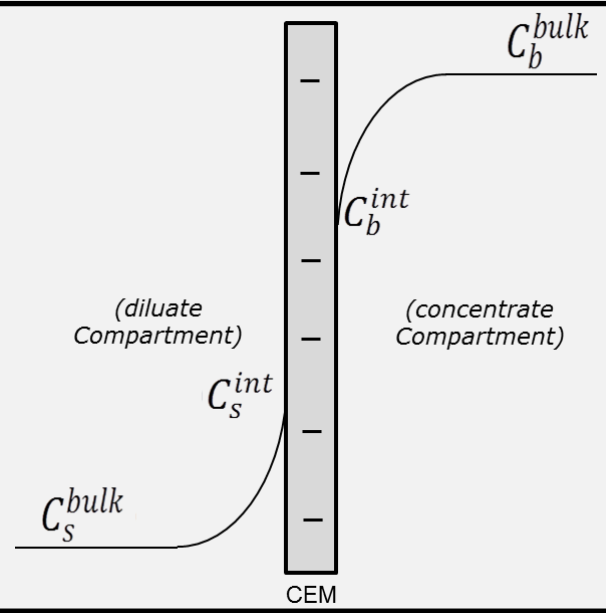


salt concentration map at **membrane-solution interface**



Polarisation Coefficients:

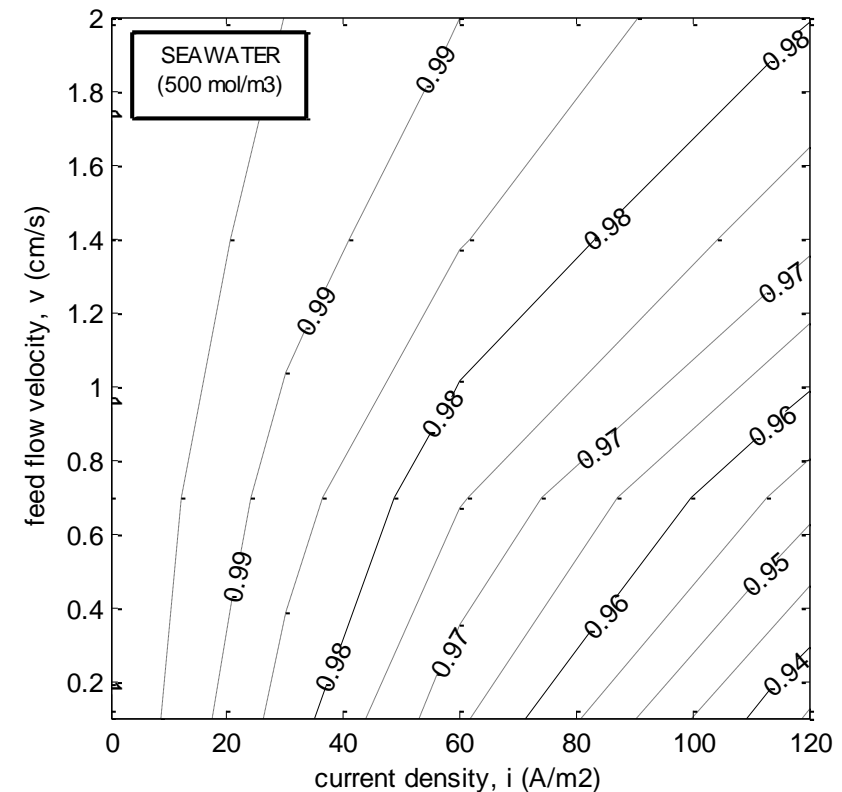
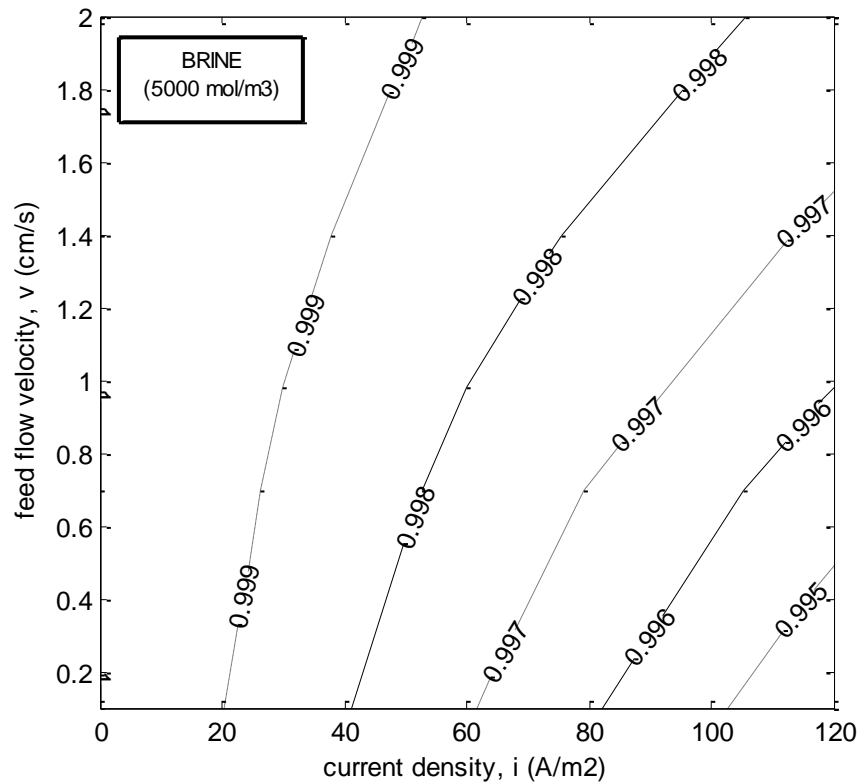
$$\left\{ \begin{aligned} \vartheta_b &= \frac{C_b^{int}}{C_b^{bulk}} \\ \vartheta_s &= \frac{C_s^{bulk}}{C_s^{int}} \end{aligned} \right.$$





# CFD Modelling: prediction of polarisation phenomena

## Polarization factor for Deukum spacer-filled channels



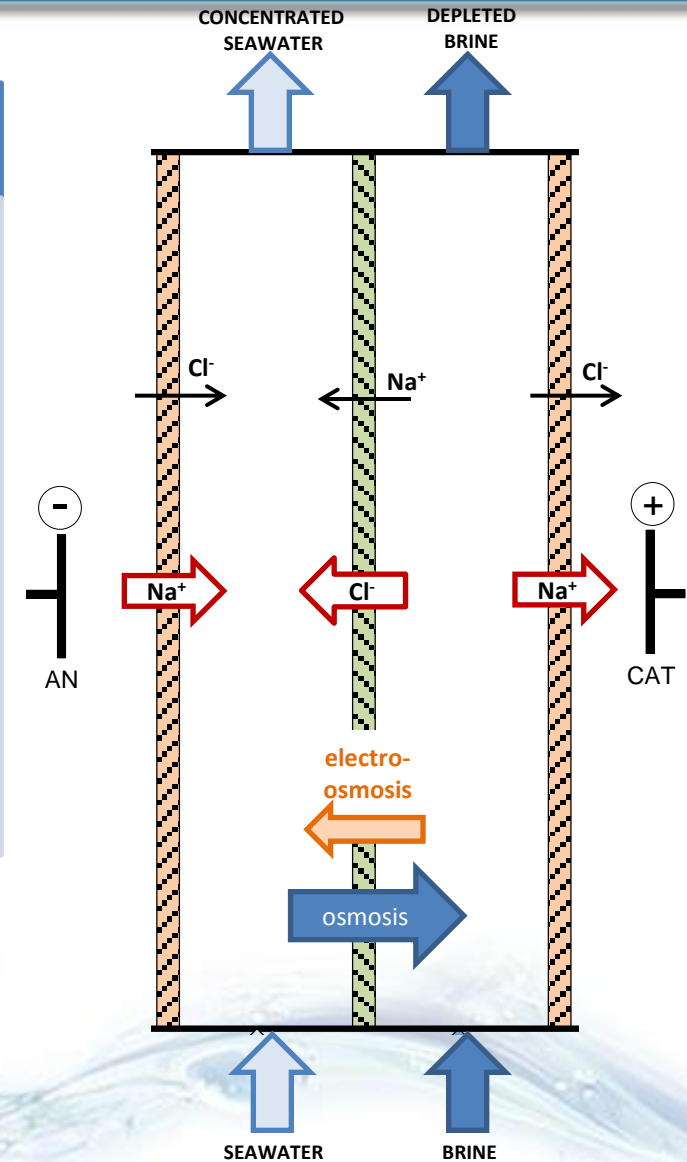
Effect of current density and fluid velocity on polarization coefficients.

CFD Model predictions of a 270  $\mu\text{m}$  polyamide woven spacer (Deukum GmbH, Germany).

# Development/validation of process simulator

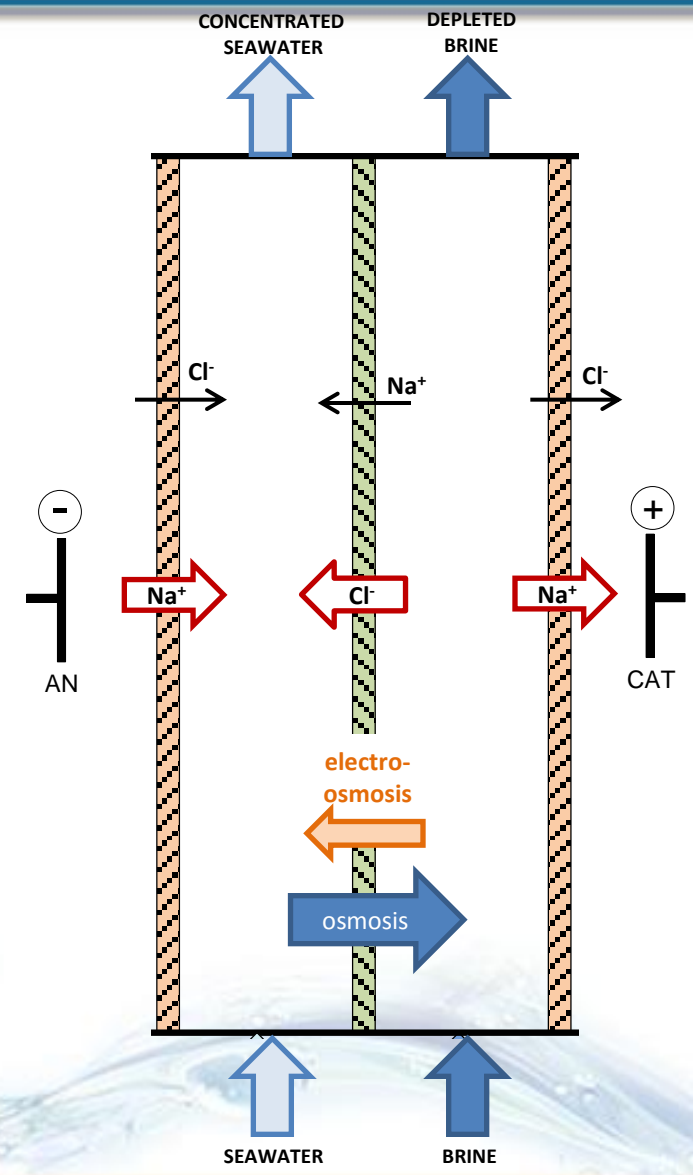
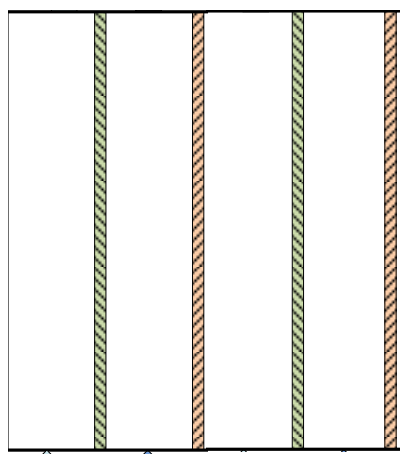
## Low-hierarchy model (*cell pair*):

- thermodynamic properties of solutions
- electric variables
- salt transport (counter/co-ions)
- solvent transport (osmosis/electro-osmosis)
- polarization phenomena
- mass balance



Source: M. Tedesco et al., *Desalination and Water Treatment*, vol. 49, pp. 404-424, 2012

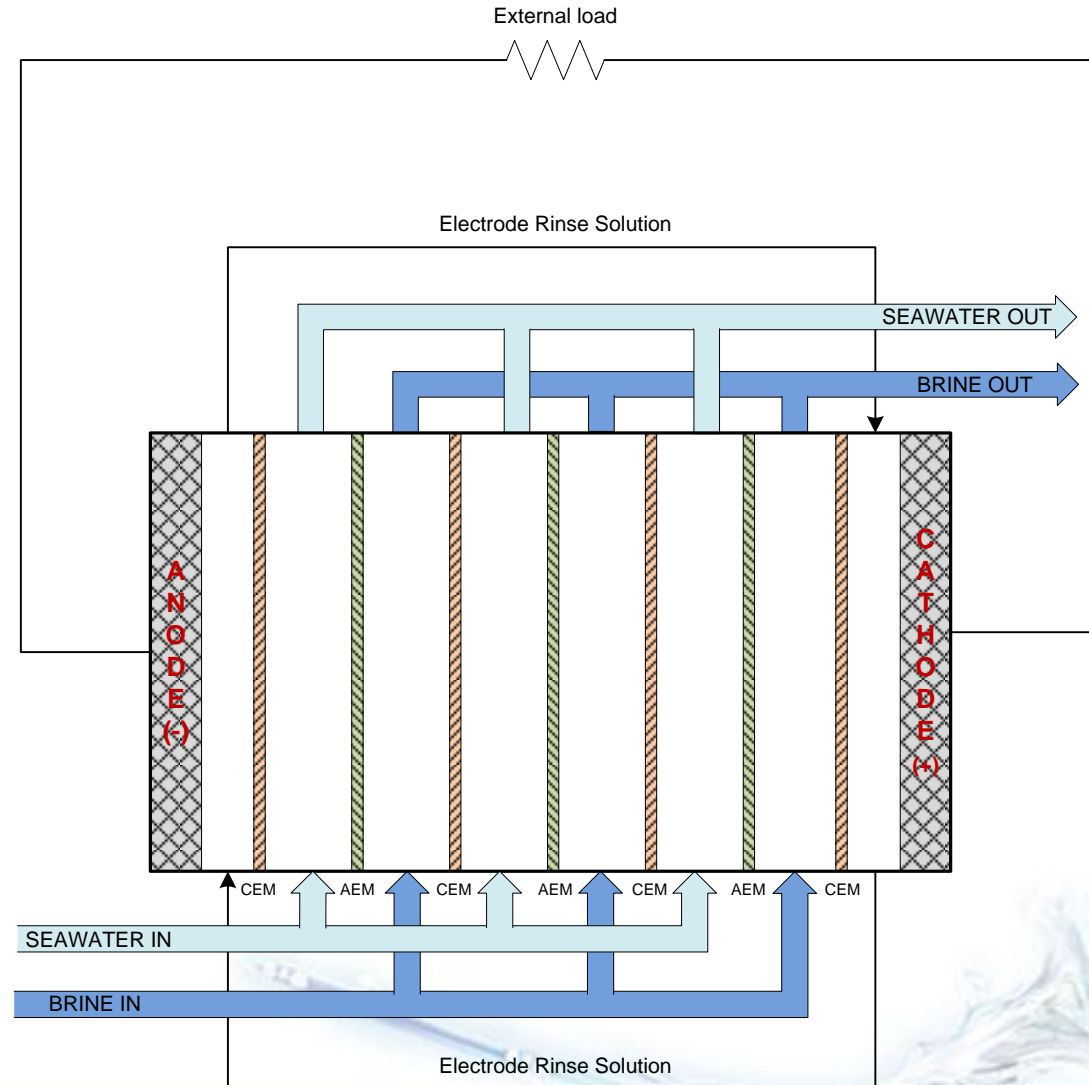
# Process Modelling Approach



Source: M. Tedesco et al., *Desalination and Water Treatment*, vol. 49, pp. 404-424, 2012



# Process Modelling Approach

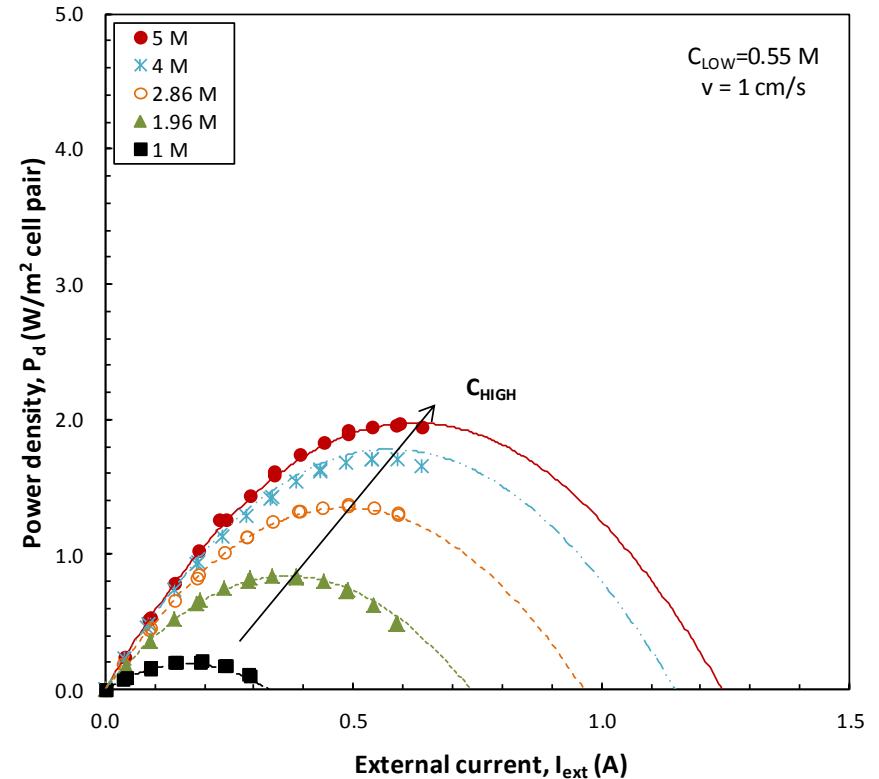
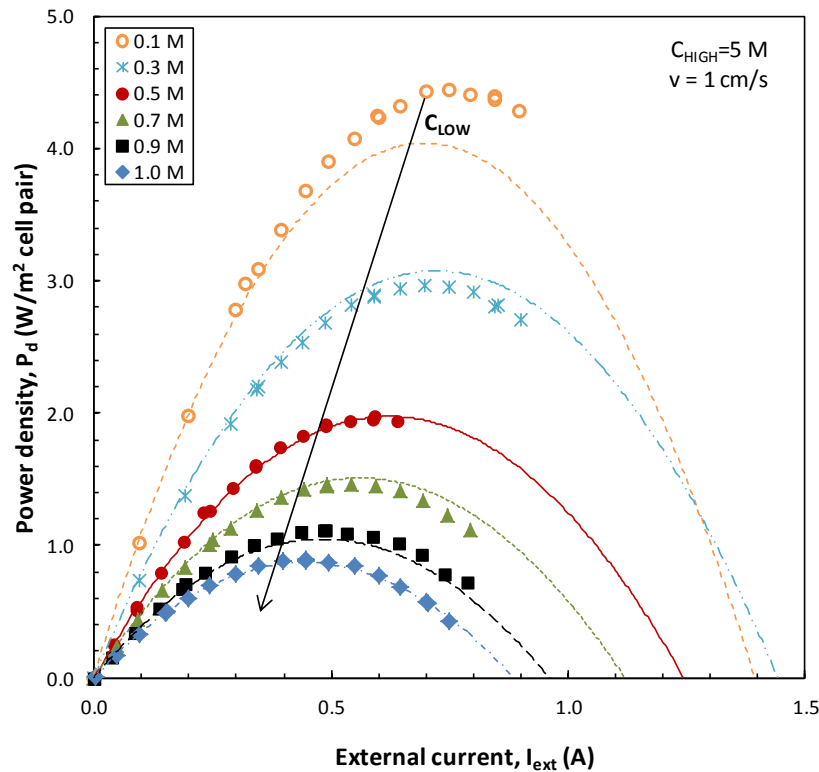


## High-hierarchy model (*stack*):

- parasitic currents through manifolds
- stack resistance
- stack voltage
- Pressure drops
- power density (gross/net)

# Process Modelling validation

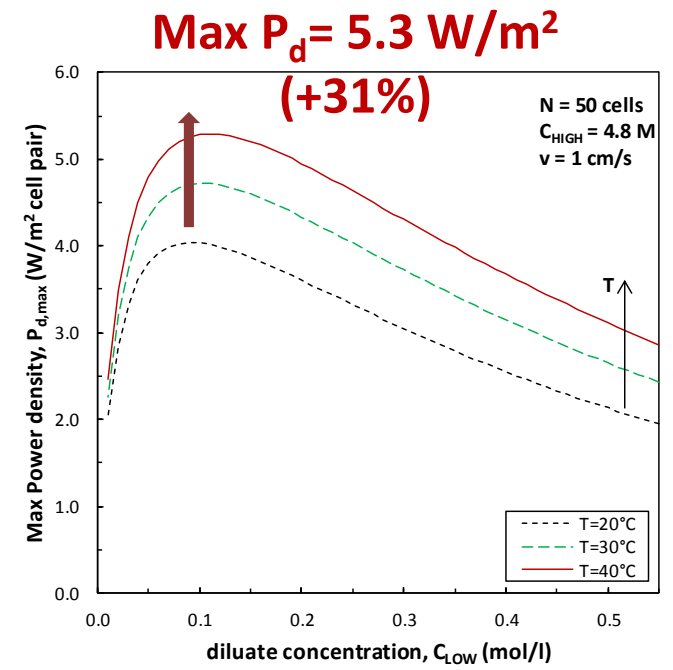
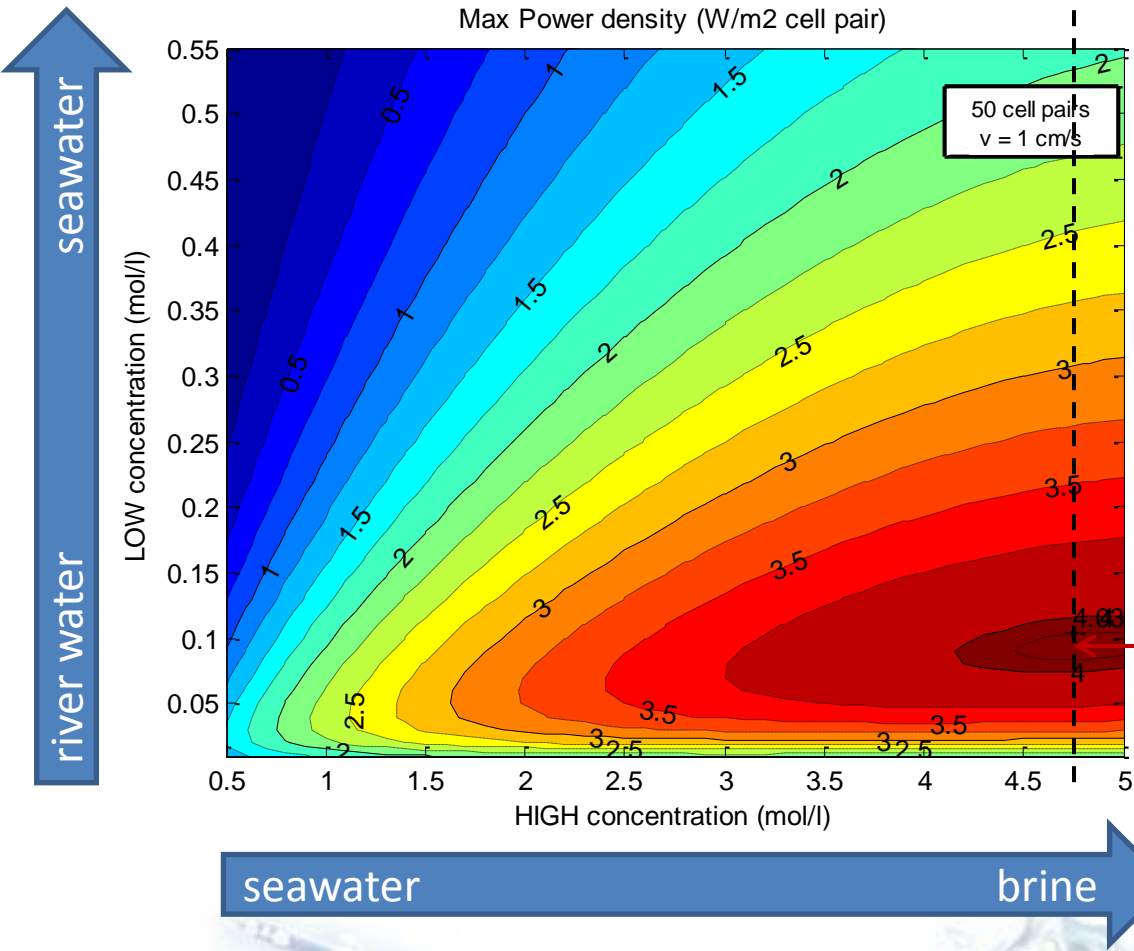
## Model calibration with variable feed concentration



Experimental (points) and simulated (lines) data for a 50-cells stack equipped with Fujifilm membranes, Deukum 270  $\mu\text{m}$  spacers; feed flow velocity: 1 cm/s;  $T=20^\circ \text{ C}$ . Blank resistance: 0.4  $\Omega$ .

# Prediction of dependences

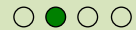
## Influence of feed T & concentration



**BEST conditions:**  
**brackish water (0.05 – 0.1 M)**  
**+ brine (4.5 – 5 M)**

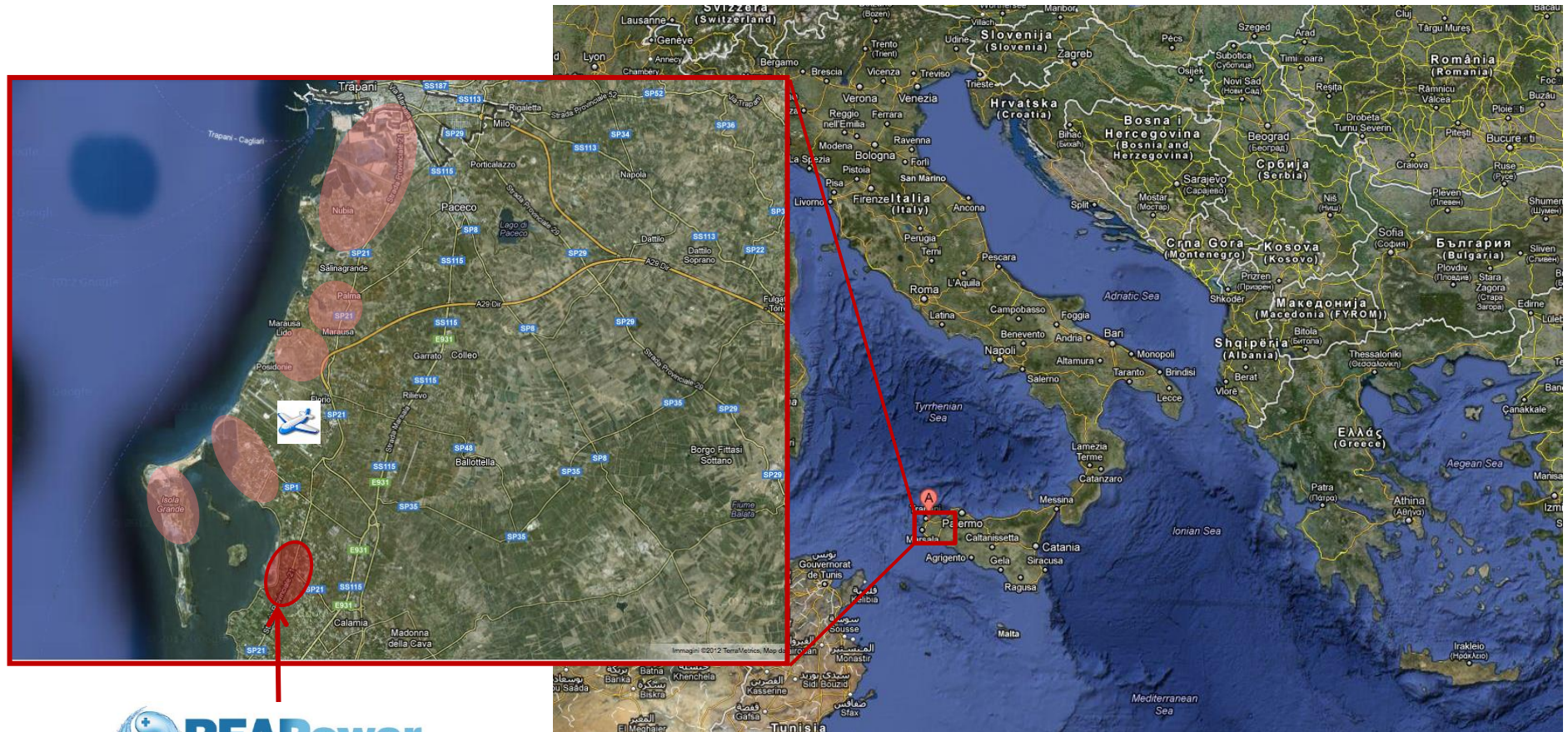
Simulations of a 50-cells stack equipped with Fujifilm membranes, Deukum spacers; fluid velocity inside channels: 1 cm/s; T=20° C. Blank resistance: 0.4 Ω.





# The REAPower prototype installation site

## The singular framework of Trapani saltworks



 **REAPower**  
**Prototype**  
**installation site**



# The REAPower prototype installation site

## The “Ettore-Infersa” saltworks

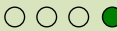


Direct access to both saturated brine and seawater from open channels

Installation place within an old, restructured WINDMILL







# Prototype installation: plant specifications

## Site features

- Seawater availability: unlimited;
- Brine availability: 10-15 m<sup>3</sup>/h (much larger with feed-recycle);
- Brine concentration: variable between 250 and 320 gr/lit.

## Provisional Prototype features

- Total cell pair surface: from 4 to 125m<sup>2</sup> (2 stacks will be tested);
- Expected power density: > 5 W/m<sup>2</sup>;
- Expected power output: from 20 to 600W

# The REAPower website



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# The Future



of sustainable energy production



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**EUROMED 2014  
PALERMO, October 2014**

***Thank you  
for your attention***

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